

NEW STANDARD ACADEMY

Date : 13-04-2026

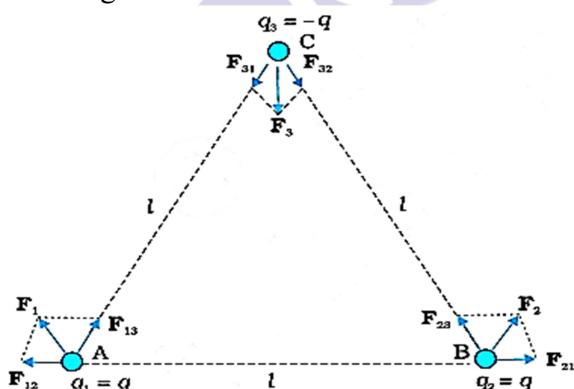
CLASS : 12TH

Marks: 60

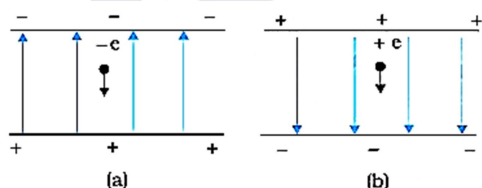
Time: 3 hours

PHYSICS

- If 10^9 electrons move out of a body to another body every second, how much time is required to get a total charge of 1 C on the other body?
- Consider the charges q , q , and $-q$ placed at the vertices of an equilateral triangle, as shown in Fig. What is the force on $-q$ charge?

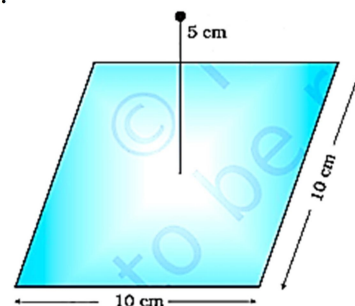


- An electron falls through a distance of 1.5 cm in a uniform electric field of magnitude $2.0 \times 10^4 \text{ N C}^{-1}$ [Fig. (a)]. The direction of the field is reversed keeping its magnitude unchanged and a proton falls through the same distance [Fig(b)]. Compute the time of fall in each case.

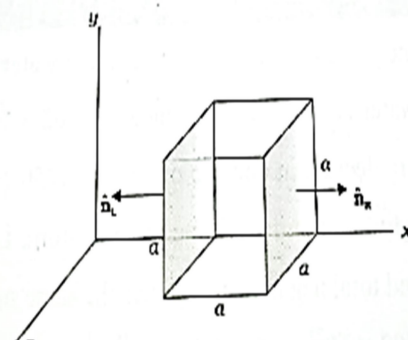


- Consider a uniform electric field $E = 3 \times 10^3 \hat{i} \text{ N/C}$. What is the flux of this field through a square of 10 cm on a side whose plane is parallel to the yz plane?
- Careful measurement of the electric field at the surface of a black box indicates that the net outward flux through the surface of the box is $8.0 \times 10^3 \text{ Nm}^2/\text{C}$. What is the net charge inside the box?
- A point charge $+10 \mu\text{C}$ is a distance 5 cm directly above the centre of a square of

side 10 cm, as shown in Fig. What is the magnitude of the electric flux through the square?



- The electric field components in Fig. are $E_x = ax^{1/2}$, $E_y = E_z = 0$, in which $a = 800 \text{ N/C m}^{1/2}$. Calculate (a) the flux through the cube, and (b) the charge within the cube. Assume that $a = 0.1 \text{ m}$.



- Two large, thin metal plates are parallel and close to each other. On their inner faces, the plates have surface charge densities of opposite signs and of magnitude $17.0 \times 10^{-22} \text{ C/m}^2$. What is E : in the outer region of the first plate?
- A point charge causes an electric flux of $-10^3 \text{ N m}^2/\text{C}$ to pass through a spherical Gaussian surface of 10.0 cm radius centred on the charge. If the radius of the Gaussian surface were doubled, how much flux would pass through the surface?
- If dielectric constant of medium is unity then what will be its permittivity?

CHEMISTRY

- E_{cell}^0 for the cell $\text{Ni}/\text{Ni}^{2+}(1\text{M})||\text{Cu}^{2+}/\text{Cu}$ is 0.59V. If $E^\circ \text{Ni}^{2+}/\text{Ni} = -0.25\text{V}$, what is E° for copper electrode?

- Potential of hydrogen electrode at one atmospheric pressure and 25°C is -0.118V. Calculate the pH of the solution.
- What is cell reaction and EMF of the following cell?

$$\text{Ni} | \text{Ni}^{2+} (0.1\text{M}) || \text{Ag}^+ (0.1\text{M}) | \text{Ag}$$
 Given: $E^\circ (\text{Ni}^{2+} | \text{Ni}) = -0.25\text{V}$,
 $E^\circ (\text{Ag}^+ | \text{Ag}) = +0.80\text{V}$
- Calculate the potential of the following cell:

$$\text{Sn}^{4+} (1.5\text{M}) + \text{Zn} \rightarrow \text{Sn}^{2+} (0.5\text{M}) + \text{Zn}^{2+} (2\text{M})$$
 Given $E^\circ (\text{Sn}^{4+} | \text{Sn}^{2+}) = 0.13\text{V}$,
 $E^\circ (\text{Zn}^{2+} | \text{Zn}) = -0.76$
 Will the cell potential increase or decrease if the concentration of Sn^{4+} is increased?
- The EMF of a cell is 0.28 V and following reaction occurs in the cell:

$$\text{Zn} + 2\text{H}^+ = \text{Zn}^{2+} (0.1\text{M}) + \text{H}_2 \text{ (one atm)}$$
 Make the cell and calculate the pH.
 $E^\circ (\text{Zn}^{2+} / \text{Zn}) = -0.76\text{V}$
- Calculate the EMF of the electrode $\text{Zn} | \text{Zn}^{2+} (0.1\text{M})$ at 298 K. Given that $E^\circ (\text{Zn}) = -0.76\text{V}$
- Calculate the EMF of $\text{Cu} | \text{CuSO}_4 (0.1\text{M})$. The salt is 90% dissociated. Given that $E^\circ (\text{Cu}^{2+} | \text{Cu}) = +0.34\text{V}$.
- Calculate the EMF of the cell at 25 °C

$$\text{Cu} | \text{Cu}^{2+} (4\text{M}) || \text{Ag}^+ (0.1\text{M}) | \text{Ag}$$
 Given that $E^\circ (\text{Cu}^{2+} / \text{Cu}) = 0.34\text{V}$ and $E^\circ (\text{Ag}^+ | \text{Ag}) = 0.80\text{V}$
- Calculate the EMF of the cell:

$$\text{Cr} | \text{Cr}^{3+} (0.1\text{M}) || \text{Fe}^{2+} (0.01\text{M}) | \text{Fe}$$
 Given that $E^\circ (\text{Cr}^{3+} | \text{Cr}) = -0.75\text{V}$; and $E^\circ (\text{Fe}^{2+} | \text{Fe}) = -0.45\text{V}$
- For the cell reaction given below EMF at 25 °C is 1.3 V.

$$\text{Zn}(\text{S}) + \text{Cu}^{2+} (1\text{M}) \rightleftharpoons \text{Cu}(\text{S}) + \text{Zn}^{2+} (0.1\text{M})$$
 Calculate E° of the cell reaction.

BIOLOGY

- What's the difference between infectious diseases and noninfectious diseases?
- What is the explain of viral diseases
- What causes pneumonia Give its Symptoms
- explain the life circle of the plasmodium.
- Why is the immune system important
- Explain the structure of HIV and labelled diagram
- Should a benign tumor be removed?
- Give difference between primary and secondary lymphoid organ.

- Define cancer.
- What is cellular barrier

MATH'S

- Find the principal value of $\cot^{-1}\left(\frac{-1}{\sqrt{3}}\right)$
- Find the value of $\cos(\sec^{-1}x + \text{cosec}^{-1}x), |x| \geq 1$
- Write $\text{Cot}^{-1}\left(\frac{1}{\sqrt{x^2-1}}\right), > 1$ in the simplest form.
- Show that
 (i) $\text{Sin}^{-1}(2x\sqrt{1-x^2}) = 2\text{sin}^{-1}x, -\frac{1}{\sqrt{2}} \leq x \leq \frac{1}{\sqrt{2}}$
 (ii) $\text{Sin}^{-1}(2x\sqrt{1-x^2}) = 2\text{cos}^{-1}x, -\frac{1}{\sqrt{2}} \leq x \leq 1$
- Prove that:- $\cot^{-1}\left(\frac{\sqrt{1+\sin x} + \sqrt{1-\sin x}}{\sqrt{1+\sin x} - \sqrt{1-\sin x}}\right) = \frac{x}{2}, x \in \left(0, \frac{\pi}{4}\right)$
- Solve the following equations
 (i) $2 \tan^{-1}(\cos x) = \tan^{-1}(2 \text{cosec } x)$
 (ii) $\tan^{-1} \frac{1-x}{1+x} = \frac{1}{2} \tan^{-1} x, (x > 0)$
- Find the value of the following:
 (i) $\cos^{-1}\left(\cos \frac{13\pi}{6}\right)$ (ii) $\tan^{-1}\left(\tan \frac{7\pi}{6}\right)$
- Solve $\tan^{-1} 2x + \tan^{-1} 3x = \frac{\pi}{400}$
- Simplify $\tan^{-1}\left[\frac{a \cos x - b \sin x}{b \cos x + a \sin x}\right], \text{ if } \frac{a}{b} \tan x > -1$
- Show that $\sin^{-1} \frac{12}{13} + \cos^{-1} \frac{4}{5} + \tan^{-1} \frac{63}{16} = \pi$